

WATER WORKS and OTHER WORDS

5th and 8th Grade (Extension Activity-1)

Go over to your sink, turn on the tap and get a big glass of cool water . . . add some ice if you like, and sit back and think on an elemental level what you are drinking. (Your tap water will contain elements and minerals essential for good health, but think back to just the basics . . . those two molecular building blocks!

You already know which elements are in water, probably learned it back in the fourth grade! (Which elements are in water? and) Hydrogen (H) and Oxygen (0) are the answers!)	ıe
You'll be hearing more in class about why there's a "subscript ₂ " after the H, and how elements combine to form compounds. But in this activity, you're going to be using the Periodic Table to locate elements by their symbolic abbreviations (symbol) or by their atomic number. What is an atomic number befine it*	∍r?
(It tells where an element ranks in weight. The smaller the number the lighter the element.)	

WHAT: you'll be locating elements to spell words and total up the atomic

number values.

HOW: look at your Periodic Table, everything you need is there! **WHEN:** in a class period or whenever your instructor tells you.

First, fill in the chart to get all the information you will need to use.

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Element	Symbol	Atomic Number
Oxygen		
	Не	
		13
Silicon		
	Н	
		6
Boron		
	N	
		73
Bismuth		
	Ва	
		16
lodine		
	U	
		24
Radium		
	Li	
		52
Tin		
	W	
		42
Carbon		
	K	

NOW, you'll be using these to spell words and add your atomic number values!

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Periodic Table of the Elements

				_	w = =		~ ~ ~ ~			_	N # #		1	
	80	2 Heium 4.002602	e e	Neon 20.1797	 >	Argon 39.948		Krypton 83.80	X &	Xenon 131.29	Rn.	Radon a (222)		
	ι	~ I ₹\$	₽ Ž		* <u>*</u>	39.₹	8구	83.	2 ×	3. K	% & C	, (22		
lectrons	ž ¥ ž	1	எட	Fluorine 18.9984032	ū	Chlorine 35.4527	ž Ž	Bromine 79.904	E	lodine 126.90447	85 At	Astatine (210)		
Number of electrons in each shell	Atomic weight () indicates atomic weight of the most	stable isotope	0	Oxygen 15.9994	<u></u>	Sulfur 32.066	Se ::	Selenium 78.96	Te	Tellurium 127.60	Po	Polonium (209)		
Num in ea	Aton () indic weight	stable 15	~ ~	Nitrogen 14.00674	~ = ~	Phosphorus 30.973762	~ V)	Arsenic 74.92159 7	~===	Antimony 121.76	****	Bismuth 5 6 208.98037 (
4	1	1	<u>, Z</u>	A 4	5 t	₹ 8	₩ ∀		2 S	4 Z	8 \(\alpha \)	4 Bis		
	Carbon 12.011	4	ູບ	Carbon 12.0107	≱ iS	Silicon 28.0855	ge Ge	Germanium 72.61	S _n	Tin 118.710	Pb	Lead 207.2		
- -	+	t t	<u>د</u> ه	Boron 10.811	ε V	Aluminum 26.981539	31 Ga	Gallium 69.723	6 디	Indium 114.818		Thallium 3 204.3833		
mbel mboł	name						~==~		~ # # *		~ * # # #	2		
Atomic number	Element name					12	[%] Z	Zinc 65.39	å D	Cadmium 112,411	#B	Mercu 200.59		
Atomic numbe Chemical symbol	Eler					Ħ	ຸກຽ	Copper 63.546	Ag :	Silver 107.8682	Au	Gold 196.96654		
						10	~ 4 4 ~		~===	ım	~==#	•		
Š	×					-	ž %	Nickel 58.6934	₽d Pd	Palladium 106.42	<u>۳</u> کے	Platinum 195.078		_
nis table gives information about the chemical elements. Elements re grouped into eight classes according to their properties. Each class shown in a different color. Hydrogen does not belong to any one class.	Actinide series	Other metals	etals	Noble gases		6	ွဒ	Cobalt 58.9332	왊	Rhodium 102.9055	77	fridium ,	Mt	(268)
nts. Eler rrties. Ea g to any	Actini	Other	Nonmetals	Nobie		. 00	26 : Fe :		Ru ::	Ruthenium 101.07	% Os ::	Osmium 2 190.23	HS :	(263)
lemel prope pelon	L						p -		~ * * * *		~ * * X X	2	~****	7.
mical e their p es not l		tals				7	Mn Mn	Manganese 54.93805		Technetium (98)	75 Re	2 Rhenium 186.207		(262)
the che rding to gen do	<u>~</u>	Alkaline earth metals	metals	Lantnanide series		9	ڻ۶	Chromium 51.9961	42 M 0	Molybdenum 95.94	⁷⁴ ≥	Tungsten 183.84	50 m	(392)
about es acco Hydro	Alkali metals	ine ea	Transition m	anide		2	M B E M	Ē	۹	Niobium 92.90638		Tantalum 2 180.9479	(A) 150 FEB 1993	
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32	8.9055	Cerium 140.116	2 Praseodymium 2 140.90765	Neodymium 2 144.24	2 Promethium 2 (145)	Samarium 2 150.36	Europium 2 151.964	Gadolinium 2 157.25	Terbium 2 158.92534	Dysprosium 162.50	m ₂ Holmium 164.93032	2 Erbium 167.26	Thullum 2 168.93421	Ytterbium 173.04	174.9
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<u> </u>	- 4	모		- M	2	= 2	= #			- 2 2	Es	F	M	2	<u>ٿ</u>
₹8	Actinium (227)	Thorium 232.0381	20 Protactinium 9 231.03588 2	238.0289 2	Neptunium 2 (237)	Plutonium 2 (244) 2	Americium 2	Curium 3		Californium	Einsteinium (252)	Fermium (257)	Mendelevium a 2 (258)	Nobelium (259)	(262)

For more information visit: http://pearl1.lanl.gov/periodic/default.htm Brought to you by the Education Partners at TIQ!



Water Works and Other Words

Say you use Oxygen's (0) and Hydrogen's (H) symbols, to spell two words and add their atomic numbers for a total POINT value of each word. You get:

OH Total points: 9 (0=8 + H=1) HOO Total points: 17 (H=1 + 0=8 + 0=8)

Let's add Boron (B) to the mix and continue on to get:

HOB Total points:14 (H=1 + 0=8 + B=5)

BOO HOO Total points:38 (B=5+0=8+0=8 and H=1+0=8+0=8)

*Use your chart to spell as many words as you can using the symbols you have noted. You can use the symbols more than once in a word! Compound words, contractions, anything goes! (But it has to be real!)

*Remember to write the atomic numbers under or next to the words to get your total point value for each word. Write that total for each word.

(Use a work sheet if that helps before you write them here!)

		be r ore you write th	
WORD	POINT VALUE	WORD	POINT VALUE
ОН	9	H O O	17
18		188	
Total Points:		Total Points:	

Could v	ou find 40 words?	GRAND Total Points (all):
COGIG ;	oa ilila to wo lasi	OKAND TOTAL TOTAL STATE.

- * Want to try some more? Use all the elements.
 - What can you come up with?
 - Longest word?
 - Most points?